

Module 4d: The Role of Pollution Prevention in Stormwater Management

Robert Pitt and Melinda Lalor
The University of Alabama at Birmingham
Birmingham, AL, USA 35294

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|---|----|
| 1.0 Introduction | 1 |
| 2.0 Sources of Urban Runoff Pollutants and Source Reduction Options..... | 1 |
| 2.1 Other Pollutant Contributions to the Storm Drainage System..... | 3 |
| 2.2 Sources of Stormwater Toxicants..... | 3 |
| 2.3 Potential Sources | 3 |
| 3.0 The Use of the Source Loading and Management Model (SLAMM) to Identify and Quantify Source Area Contributions | 5 |
| 4.0 Potential Materials and Alternatives Available for Stormwater Pollution Prevention..... | 6 |
| 4.1 Roofing and Paving Materials | 7 |
| 4.2 Exposed Wooden Material/Treated Wood..... | 8 |
| 5.0 Leaching of Various Construction Materials..... | 9 |
| 6.0 Conclusions | 11 |
| 7.0 References | 12 |

1.0 Introduction

Around the nation, there is growing interest in the development and use of environmentally sensitive construction materials as a low-cost component to stormwater management. It is thought that the more appropriate selection of materials that are exposed to the environment should result in significant reductions of many toxicants in stormwater. Unfortunately, there is little data for specific alternative building materials, although much information exists targeting selected sources, especially the role of roof runoff as a significant source of zinc and other metals.

Past studies have identified urban runoff as a major contributor to the degradation of many urban streams and rivers (such as Field and Turkeltaub 1981; Pitt and Bozeman 1982; Pitt and Bissonnette 1984; Pitt 1995). Previous studies also found organic and metallic toxicants in urban storm-induced discharges that can contribute to receiving water degradation (such as EPA 1983; Hoffman, *et al.* 1984; Fram, *et al.* 1987). Studies conducted by Pitt, *et al.* (1995 and 2000) investigated toxic contributions to urban wet weather flow from sources such as roofs, parking areas, storage areas, streets, loading docks, vehicle service areas, and landscaped areas. Roof, vehicle service area and parking lot runoff samples were found to have the greatest organic toxicant detection frequencies and the highest levels of detected metals. Research is currently underway at UAB to develop effective procedures for treating runoff from vehicle service areas and parking lots at its source (Clark and Pitt 1999; Pitt, *et al.*, 2000). These areas are particularly subject to spills and leaks of automotive products and exhaust emissions from frequently starting vehicles. These areas are usually isolated enough to make source area runoff treatment feasible. However, relative pollutant contributions from various roofing, wooden and paving materials themselves are also a concern which has not been adequately addressed. Due to the common use of these surfaces in our urban environments, reduction of emissions at the source is desirable, and material substitution would seem a good place to start.

2.0 Sources of Urban Runoff Pollutants and Source Reduction Options

It has been known for many years that the vast majority of stormwater toxicants and much of the conventional pollutants are associated with automobile use and maintenance activities and that these pollutants are strongly associated with the particulates suspended in the stormwater (the non-filterable components, or suspended solids). It has been difficult to reduce or modify automobile use to reduce the use of these compounds, with the notable

exception of the phasing out of leaded gasoline. Current activities, concentrated in the San Francisco area, are trying to encourage brake pad manufactures to reduce the use of copper. The effectiveness of most stormwater control practices is therefore dependent on their ability to remove these particles from the water, or possibly from intermediate accumulating locations (such as streets or other surfaces) and not through source reduction. The removal of these particles from stormwater is dependent on various characteristics of these particles, especially their size and settling rates. Some source area controls (most notably street cleaning) affect the particles before they are washed-off and transported by the runoff, while others remove the particles from the flowing water.

Table 1 shows that most of the organic compounds found in stormwater are associated with various human-related activities, especially automobile and pesticide use, or are associated with plastics (Verschueren 1983). Heavy metals found in stormwater also mostly originate from automobile use activities, including gasoline combustion, brake lining, fluids (brake fluid, transmission oil, anti-freeze, grease, etc.), undercoatings, and tire wear (Durum 1974, Koeppe 1977, Rubin 1976, Shaheen 1975, Solomon and Natusch 1977, and Wilbur and Hunter 1980). Auto repair, pavement wear, and deicing compound use also contribute heavy metals to stormwater (Field, *et al.* 1973, and Shaheen 1975). Shaheen (1975) found that eroding area soils are the major source of the particulates in stormwater. The eroding area soil particles, and the particles associated with road surface wear, become contaminated with exhaust emissions and runoff containing the polluting compounds. Most of these compounds become tightly bound to these particles and are then transported through the urban area and drainage system (or removed) with the particulates. Stormwater concentrations of zinc, fluoranthene, 1,3-dichlorobenzene, and pyrene are unique in that substantial fractions of these compounds remain in the water and are less associated with the particulates.

Table 1. Uses and Sources for Organic Compounds found in Stormwater (Source: Verschueren 1983)

| Compound | Example Use/Source |
|----------------------------|---|
| Phenol | gasoline, exhaust |
| N-Nitroso-di-n-propylamine | contaminant of herbicide Treflan |
| Hexachloroethane | plasticizer in cellulose esters, minor use in rubber and insecticide |
| Nitrobenzene | solvent, rubber, lubricants |
| 2,4-Dimethylphenol | asphalt, fuel, plastics, pesticides |
| Hexachlorobutadiene | rubber and polymer solvent, transformer and hydraulic oil |
| 4-Chloro-3-methylphenol | germicide; preservative for glues, gums, inks, textile, and leather |
| Pentachlorophenol | insecticide, algacide, herbicide, & fungicide mfg., wood preservative |
| Fluoranthene | gasoline, motor and lubricating oil, wood preservative |
| Pyrene | gasoline, asphalt, wood preservative, motor oil |
| Di-n-octylphthalate | general use of plastics |

All areas are affected by atmospheric deposition, while other sources of pollutants are specific to the activities conducted on the areas. As examples, the ground surfaces of unpaved equipment or material storage areas can become contaminated by spills and debris, while undeveloped land remaining relatively unspoiled by activities can still contribute runoff solids, organics, and nutrients, if eroded. Atmospheric deposition, deposition from activities on paved surfaces, and the erosion of material from upland unconnected areas are the major sources of pollutants in urban areas.

The important sources of these pollutants are related to various uses and processes. Automobile related potential sources usually affect road dust and dirt quality more importantly than other particulate components of the runoff system. The road dust and dirt quality is affected by vehicle fluid drips and spills (gasoline, oils, etc.) and vehicle exhaust, along with various vehicle wear, local soil erosion, and pavement wear products. Urban landscaping practices potentially affecting urban runoff include vegetation litter, fertilizer and pesticide. Miscellaneous sources of urban runoff pollutants include firework debris, wildlife and domestic pet wastes and possibly industrial and sanitary wastewaters. Wet and dry atmospheric contributions both affect runoff quality. Pesticide use in an urban area can contribute significant quantities of various toxic materials to urban runoff. Many manufacturing and industrial activities, including the combustion of fuels, also affects urban runoff quality.

Natural weathering and erosion products of rocks contribute the majority of the hardness and iron in urban runoff pollutants. Road dust and associated automobile use activities (gasoline exhaust products) historically contributed

most of the lead in urban runoff. However, the decrease of lead in gasoline has resulted in current stormwater lead concentrations being about 1/10 of the levels found in stormwater in the early 1970s (Bannerman, *et al.* 1993). In certain situations, paint chipping can also be a major source of lead in urban areas. Road dust contaminated by tire wear products, and zinc plated metal erosion material, contribute most of the zinc to urban runoff. Urban landscaping activities can be a major source of cadmium (Phillips and Russo 1978). Electroplating and ore processing activities can also contribute chromium and cadmium.

Many pollutant sources are specific to a particular area and on-going activities. For example, iron oxides are associated with welding operations and strontium, used in the production of flares and fireworks, would probably be found on the streets in greater quantities around holidays, or at the scenes of traffic accidents. The relative contribution of each of these potential urban runoff sources, is, therefore, highly variable, depending upon specific site conditions and seasons.

2.1 Other Pollutant Contributions to the Storm Drainage System

The detection of pentachlorophenols in stormwater indicates leaching from treated wood. Frequent detections of polycyclic aromatic hydrocarbons (PAHs) during the U.S. Environmental Protection Agency’s Nationwide Urban Runoff Program (EPA 1983) may possibly indicate leaching from creosote treated wood, in addition to fossil fuel combustion sources. High concentrations of copper, and some chromium and arsenic observations also indicate the potential of leaching from “CCA” (copper, chromium, and arsenic) treated wood. The significance of these leachate products in the receiving waters is currently unknown, but alternatives to these preservatives should be considered. Many cities use aluminum and concrete utility poles instead of treated wood poles. This is especially important considering that utility poles are usually located very close to the drainage system ensuring an efficient delivery of leachate products. Many homes currently use wood stains containing pentachlorophenol and other wood preservatives. Similarly, the construction of retaining walls, wood decks and playground equipment with treated wood is common. Some preservatives (especially creosote) cause direct skin irritation, besides contributing to potential problems in receiving waters. Many of these wood products are at least located some distance from the storm drainage system, allowing some improvement to surface water quality by infiltration through pervious surfaces.

2.2 Sources of Stormwater Toxicants

Tables 2 and 3 summarize toxicant concentrations and likely sources or locations having some of the highest concentrations found by Pitt, *et al.* (1995). The detection frequencies for the heavy metals are all close to 100 percent for all source areas, while the detection frequencies for the organics shown ranged from about 10 to 25 percent. Vehicle service areas had the greatest abundance of observed organics, with landscaped areas having many of the observed organics.

Table 2. Heavy Metal Source Area Observations (Pitt, *et al.* 1995)

| Toxicant | Highest median conc. (µg/L) | Source Area | Highest conc. (µg/L) | Source Area |
|----------|-----------------------------|-----------------------------|----------------------|------------------------|
| Cadmium | 8 | vehicle service area runoff | 220 | street runoff |
| Chromium | 100 | landscaped area runoff | 510 | roof runoff |
| Copper | 160 | urban receiving water | 1250 | street runoff |
| Lead | 75 | CSO | 330 | storage area runoff |
| Nickel | 40 | parking area runoff | 130 | landscaped area runoff |
| Zinc | 100 | roof runoff | 1580 | roof runoff |

2.3 Potential Sources

A drainage system captures runoff and pollutants from many source areas, all with individual characteristics influencing the quantity of runoff and pollutant load. Impervious source areas may contribute most of the runoff during small storm events (e.g., paved parking lots, streets, driveways, roofs, sidewalks, etc.). Pervious source areas can have higher material washoff potentials and become important contributors for larger storm events when their

infiltration rate capacity is exceeded (e.g., gardens, bare ground, unpaved parking areas, construction sites, undeveloped areas, etc.). Many other factors also affect the pollutant contributions from source areas, including: surface roughness, vegetative cover, gradient, and hydraulic connections to a drainage system; rainfall intensity, duration, and antecedent dry period; and pollutant availability due to direct contamination from local activities, cleaning frequency/efficiency, and natural and regional sources of pollutants. The relative importance of the different source areas is therefore a function of the area characteristics, pollutant washoff potential, and the rainfall characteristics (Pitt 1987).

Table 3. Toxic Organic Source Area Observations (Pitt, *et al.* 1995)

| Toxicant | Maximum ($\mu\text{g/L}$) | Detection Frequency (%) | Significant Sources |
|-------------------------------|--------------------------------|----------------------------|---|
| Benzo (a) anthracene | 60 | 12 | gasoline, wood preservative |
| Benzo (b) fluoranthene | 226 | 17 | gasoline, motor oils |
| Benzo (k) fluoranthene | 221 | 17 | gasoline, bitumen, oils |
| Benzo (a) pyrene | 300 | 17 | asphalt, gasoline, oils |
| Fluoranthene | 128 | 23 | oils, gasoline, wood preservative |
| Naphthalene | 296 | 13 | coal tar, gasoline, insecticides |
| Phenanthrene | 69 | 10 | oils, gasoline, coal tar |
| Pyrene | 102 | 19 | oils, gasoline, bitumen, coal tar, wood preservative |
| Chlordane | 2.2 | 13 | insecticide |
| Butyl benzyl phthalate | 128 | 12 | plasticizer |
| Bis (2-chloroethyl) ether | 204 | 14 | fumigant, solvents, insecticides, paints, lacquers, varnishes |
| Bis (2-chloroisopropyl) ether | 217 | 14 | pesticides |
| 1,3-Dichlorobenzene | 120 | 23 | pesticides |

Important sources of toxicants are often related to the land use (e.g., high traffic capacity roads, industrial processes, and storage area) that are unique to specific land uses activities. Automobile related sources affect the quality and quantity of road dust particles through gasoline and oil drips/spills; deposition of exhaust products; and wear of tire, brake, and pavement materials (Shaheen 1975). Urban landscaping practices potentially produce vegetation cuttings and fertilizer and pesticide washoff. Miscellaneous sources include holiday firework debris, wildlife and domestic pet wastes, and possible sanitary wastewater infiltration. In addition, resuspension and deposition of pollutants/particles via the atmosphere can increase or decrease the contribution potential of a source area (Pitt and Bozeman 1982; Bannerman, *et al.* 1993).

Numerous source area samples were collected by Pitt, *et al.* (1995). Roof runoff, parking area and vehicle service area samples had the greatest detection frequencies for the organic toxicants. Vehicle service areas and urban creeks had several of the observed maximum organic compound concentrations. Most of the organics were associated with the non-filtered sample portions, indicating an association with the particulate sample fractions. The compound 1,3-dichlorobenzene was an exception, having a significant dissolved fraction.

In contrast to the organics, the heavy metals analyzed were detected in almost all samples, including the filtered sample portions. The non-filtered samples generally had much higher concentrations, with the exception of zinc which was mostly associated with the dissolved sample portion (i.e., not associated with the SS). Roof runoff generally had the highest concentrations of zinc, probably from galvanized roof drainage components, as previously reported by Bannerman, *et al.* (1983), and others. Parking and storage areas had the highest nickel concentrations, while vehicle service areas and street runoff had the highest concentrations of cadmium and lead. Urban creek samples had the highest copper concentrations, which were probably due to illicit industrial connections or other non-stormwater discharges.

Table 4 shows the relative toxicities of the collected stormwaters. A wide range of toxicities were found. About 9% of the non-filtered samples were considered highly toxic using the Microtox™ toxicity screening procedure. About 32% of the samples were moderately toxic and about 59% were considered non-toxic. The greatest percentage of samples considered the most toxic were from industrial storage and parking areas. Landscaped areas also had a high

incidence of highly toxic samples (presumably due to landscaping chemicals), and roof runoff had some highly toxic samples (presumably due to high zinc concentrations). The chemical analyses also generally found much higher toxicant concentrations in the non-filtered sample portions, compared to the filtered sample portions.

Table 4. Relative Toxicity of Samples Using Microtox™ (Non-filtered) (Pitt, et al. 1995)

| Local Source Areas | Highly Toxic (%) | Moderately Toxic (%) | Not Toxic (%) | Number of Samples |
|-----------------------|------------------|----------------------|---------------|-------------------|
| Roofs | 8 | 58 | 33 | 12 |
| Parking Areas | 19 | 31 | 50 | 16 |
| Storage Areas | 25 | 50 | 25 | 8 |
| Streets | 0 | 67 | 33 | 6 |
| Loading Docks | 0 | 67 | 33 | 3 |
| Vehicle Service Areas | 0 | 40 | 60 | 5 |
| Landscaped Areas | 17 | 17 | 66 | 6 |
| Urban Creeks | 0 | 11 | 89 | 19 |
| Detention Ponds | 8 | 8 | 84 | 12 |
| All Areas | 9% | 32% | 59% | 87 |

Microbics suggested toxicity definitions for 35 minute exposures:

Highly Toxic - light decrease >60%

Moderately Toxic - light decrease <60% & >20%

Not Toxic - light decrease <20%

3.0 The Use of the Source Loading and Management Model (SLAMM) to Identify and Quantify Source Area Contributions

SLAMM, the Source Loading and Management Model, was originally developed to better understand the relationships between sources of urban runoff pollutants and runoff quality (Pitt and Voorhees 1996). It has been continually expanded since the late 1970s and now includes a wide variety of source area and outfall control practices (infiltration practices, wet detention ponds, porous pavement, street cleaning, catchbasin cleaning, and grass swales). SLAMM is strongly based on actual field observations, with minimal reliance on pure theoretical processes that have not been adequately documented or confirmed in the field. SLAMM is mostly used as a planning tool, to better understand sources of urban runoff pollutants and their control.

Special emphasis has been placed on small storm hydrology and particulate washoff in SLAMM. Many currently available urban runoff models have their roots in drainage design where the emphasis is with very large and rare rains. In contrast, stormwater quality problems are mostly associated with common and relatively small rains. The assumptions and simplifications that are legitimately used with drainage design models are not appropriate for water quality models. SLAMM therefore incorporates unique process descriptions to more accurately predict the sources of runoff pollutants and flows for the storms of most interest in stormwater quality analyses. However, SLAMM can be effectively used in conjunction with drainage design models to incorporate the mutual benefits of water quality controls on drainage design. Many SLAMM user's have also incorporated the use of the model with a GIS.

The development of SLAMM began in the mid 1970s, primarily as a data reduction tool for use in early street cleaning and pollutant source identification projects sponsored by the EPA's Storm and Combined Sewer Pollution Control Program, and has been greatly expanded over the past 20+ years. SLAMM can now be effectively used as a tool to enable watershed planners to obtain a better understanding sources of pollutants and of the effectiveness of different control practice programs. Various attributes of SLAMM have been published in Volumes 6 through 8 of the proceedings of the stormwater user's conference given annually in Toronto (Pitt 1997; Pitt 1998; Pitt and Lantrip 1999).

One of the first problems in evaluating an urban area for stormwater controls is the need to understand where the pollutants of concern are originating under different rain conditions. Figure 1 is an example for a typical medium density residential area showing the percentage of stormwater volume originating from different major source areas, as a function of rain depth. For storms of up to about 0.1 inch in depth, street surfaces contribute about one-half to

the total runoff to the outfall. This contribution decreased to about 20 percent for storms greater than about 0.25 inch in depth. This decrease in the significance of streets as a source of stormwater is associated with an increase of water contributions from landscaped areas (which make up more than 75% of the area and have clayey soils). Similarly, the significance of runoff from driveways and roofs also starts off relatively high and then decreases with increasing storm depth. Obviously, this is just an example and the source contributions would vary greatly for different land uses/development conditions, rainfall patterns, and the use of different source area controls.

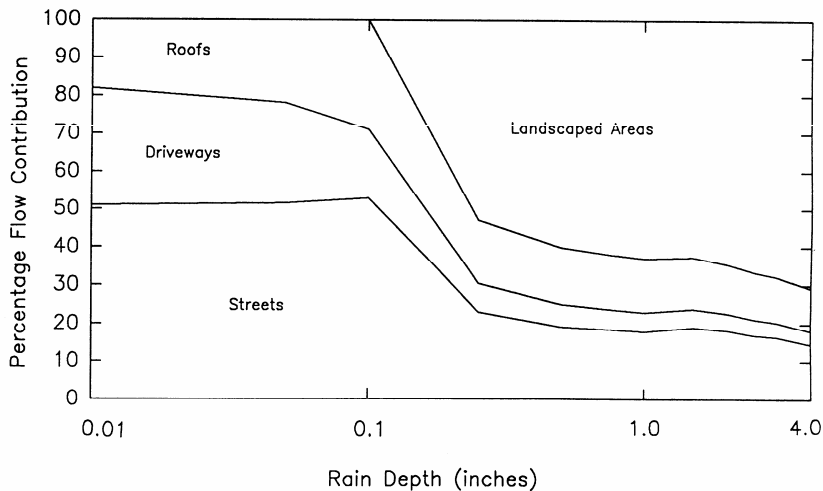


Figure 1. Flow sources for example medium density residential area having clayey soils (Pitt and Voorhees 1996).

SLAMM can be used to quantify the effects that different urban surfaces have on stormwater quality. In the above example, the roofs contributed about 10 to 20 percent of the runoff water (typical for most residential areas), but additional analyses indicate that roof runoff contributes the majority of the zinc in stormwater. The importance of pavements is also illustrated when using SLAMM for most areas, showing a potential for pollutant discharge modifications if alternative materials are used. The following section is a review of stormwater contributions from different roofing and paving materials. It is clear that pollution prevention opportunities are available with the careful selection of materials used in construction.

4.0 Potential Materials and Alternatives Available for Stormwater Pollution Prevention

Based on the results of many source area monitoring activities, candidate urban surfaces having potential for pollution reductions through the appropriate selection of alternative materials include roofing and paving materials. Building siding is also of concern as it is also exposed to rain and may cause some of the same problems currently found for roofing. The use of treated wood is also a concern. The following list shows typical components used for roofs and pavement surfaces:

For Roofing Materials

- concrete (roofing tiles)
- glass (sky lights)
- clay (roofing tiles)
- tar (flat roofing material)
- gravel (flat roofing material)
- asphalt/asbestos (roofing shingles)
- wood (roofing shingles and shakes)

- zinc (flashing and roofing panels)
- copper (flashing, gutter and roofing panels)
- aluminum (flashing, gutter, and drain material)
- galvanized metal (flashing, gutter, and drain material)
- plastic/rubber (membrane roofing)
- roofing felt (under shingles)
- roofing nails
- plastic glue/mastic (patching compound)
- PVC plastic (gutter and drain material)

For Paved Surfaces:

- Asphaltic cement flexible pavement
- Portland cement rigid pavement

4.1 Roofing and Paving Materials

Boller (1997) identified heavy metals such as cadmium, copper, lead and zinc as the critical metals in domestic wastewaters and, based on his flow studies, concluded that runoff from roofs and streets contribute 50-80% of these metals to the total mass flow in Swiss combined sewer systems. Roof runoff samples, from tile, polyester, and flat gravel roofs, were analyzed and metal concentrations were found to vary tremendously with roof type. First flush analyses showed polyester roofs contributing highest concentrations of copper (6,817 µg/L), zinc (2,076 µg/L), cadmium (3.1 µg/L) and lead (510 µg/L). Concentrations in runoff from tile roofs were copper (1,905 µg/L), zinc (360 µg/L), cadmium (2.1 µg/L) and lead (172 µg/L). Runoff from flat gravel roofs also contributed copper (140 µg/L), zinc (36 µg/L), cadmium (0.2 µg/L) and lead (22 µg/L). Runoff from roofs was found to contain not only heavy metals, but polycyclic aromatic hydrocarbons (PAHs) and organic halogens as well.

Mottier and Boller (1996), working in Zurich, measured metal concentrations in road runoff and found average values of 300 µg/L for lead, 4 µg/L for cadmium, 150 µg/L for copper and 500 µg/L for zinc. Information on pavement material type was not included. Averaged roof runoff concentrations (from tile and polyester roofs) were also measured at 16 µg/L for lead, 0.17 µg/L for cadmium, 225 µg/L for copper and 42 µg/L for zinc. Boller concluded that copper installations on buildings seem to represent the largest source for the emission of this metal into the environment. Stark, *et al.* (1995) arrived at a similar conclusion, estimating that stormwater from roofs may be responsible for more the 60% of the copper in Austria's combined sewers.

Researchers in Marquette, Michigan, collecting wet weather flow concurrently at 33 sites during 12 storms detected discernable differences in runoff quality between a variety of impervious source areas. Commercial and residential rooftops were found to produce the lowest concentration of suspended solids, but the highest concentration of dissolved metals such as lead, zinc, cadmium, and copper. Parking lots produced the highest concentrations for all PAH compounds and high concentrations of zinc, total cadmium and total copper. Low traffic streets were also identified as a major producer of total cadmium (Steuer, *et al.* 1997).

Jurgen Forster (1996) sampled and analyzed roof runoff for heavy metals (Cd, Cu, Zn, Pb) between April 1993 and May 1994. Measurement were made with an experimental roof system situated on the Campus of the University of Bayreuth and at various locations in the urban area of Bayreuth, Northern Bavaria. The experimental roof systems allowed the influence of different roof materials (concrete tiles, zinc sheet, pantiles, fibrous cement) on runoff quality to be compared. Large differences in runoff pollutant concentrations from various roofs were interpreted to indicate that the pollutants were not only being transported to the surface via the atmosphere, but also originating from the material itself. Extremely high values of zinc and copper were measured when the roof system, or parts of it, were made of metal panels, flashing and gutters. For example, runoff concentrations from zinc sheet roofing started almost three orders of magnitude higher and remained more than twenty times above the values measured for the roofs affected only by atmospheric deposition. Forster noted the most critical effect of runoff pollution containing heavy metals is their high ecotoxicity in receiving waters. Mean runoff concentration values at his study sites exceeded by about two orders of magnitude local toxicity thresholds Peak values exceeded thresholds by a

factor of 1000 or more. Forster concluded by advocating abandoning the use of exposed metal surfaces on roofs and walls of buildings.

Good (1993) reported the results of one time sampling of runoff from a rusty galvanized metal roof, a weathered metal roof, a built-up roof of plywood covered with roofing paper and tar, a flat tar-covered roof which had been painted with a fibrous reflective aluminum paint, and a relatively new anodized aluminum material at a sawmill facility on the coast of Washington. The research was carried out following the discovery that stormwater samples from the site were acutely toxic and contained high concentrations of zinc. Differences in contributions of copper, lead, and zinc were noticed between each roof type. Built-up roofing contributed the highest concentrations of dissolved copper (128 µg/L) and total copper (166 µg/L), approximately 10 times higher than levels detected in runoff from the other roofs sampled. Runoff from the rusty galvanized metal roof contained the highest concentrations of dissolved lead (35 µg/L) and total lead (302 µg/L), dissolved zinc (11,900 µg/L) and total zinc (12,200 µg/L). High concentrations of zinc were noted in runoff from each type of roof sampled at the site. Dissolved metals concentrations and toxicity remained high in roof runoff samples collected three hours after the beginning of the storm event, indicating metals leaching continued throughout storm events. All roof runoff samples were found to be highly toxic to rainbow trout, with the aluminum painted roof least toxic. Roof runoff sample concentrations exceeded the water quality criteria for copper, lead, and zinc in all samples, though the greatest exceedences were for zinc. Acid rain and the high ionic content of the coastal atmosphere were thought to have contributed to the rapid corrosion of the galvanized metal roofs and leaching of zinc. Interestingly, plastic rain gutters were reported as a source of lead.

Gumbs and Dierberg (1985) also cited the corrosion of galvanized roofs in a coastal environment as a source of heavy metal pollution. Yaziz, *et al.* (1989) analyzed the zinc content of roof runoff during rainfall events in Malaysia and observed continued elevated zinc levels in roof runoff after the first flush, indicating that zinc was leaching from the galvanized roof surface during the storm.

Thomas and Greene (1993) working in and near Armidale, Australia, found differences in metal contaminate levels between urban and rural roofs associated with variations in atmospheric deposition and differences related to antecedent dry periods. He also found runoff water quality influenced by different roof types. Zinc concentrations were significantly higher in galvanized iron roof catchments, while pH, conductivity and turbidity levels were higher in concrete tile roof catchments.

Pitt, *et al.* (1995) found high concentrations of organic constituents in runoff from several types of paved source areas. Paved areas receive pollutant contributions from vehicle exhaust emissions, tire and brake wear, vehicle corrosion and leaks, carry-in and atmospheric deposition, which are then washed off to varying degrees in subsequent rains. However, differences noted between sampling sites indicate potential differences in contribution of organics from paving materials themselves. Polycyclic aromatic hydrocarbons (PAHs), in particular, are of concern, because they are known to have potential for adverse effects to a large number of invertebrates, fishes, birds, and mammals (Kennish, 1992). Chlorination of PAHs in water treatment plants have also been found to produce carcinogenic by-products (Kopfler, *et al.* 1977).

4.2 Exposed Wooden Material/Treated Wood

The literature also supports the concern of toxicant leaching potential associated with a variety of woods, especially treated woods, used for utility poles, recreational and other wooden structures. Typical treated woods include chromated-copper-arsenate (CCA), ammoniacal copper zinc arsenate (ACZA), pentachlorophenol (PCP), and creosote. The volume of treated wood produced in the United States in 1987 was as follows: CCA/ACZA – 11.9 million cubic meters, PCP – 1.4 million cubic meters, Creosote – 2.8 million cubic meters (Micklewright 1989).

Both arsenic and chromium are heavy metals which have acute environmental health risks associated with them. Copper does not generally constitute a human health risk, however, low concentrations of copper, in certain ionic forms, are highly toxic to marine fauna and flora. The known toxicity of arsenic and chromium to humans has resulted in concern about the possible introduction into the environment of large amounts of these metals in treated wood products (Brooks 1993).

Pentachlorophenol is a highly chlorinated, synthetic preservative containing pentachlorophenol, 2,3,4,6-tetrachlorophenol, higher chlorophenols, dioxins and furans (Shields, *et al.* 1976). Arsenault (1975) and Stranks (1976) reported the presence of pentachlorophenol around the base, and in drainage ditches near treated utility poles. Stranks reported drainage ditch waters with 1.8 times the 96-h LC50 of chlorophenol for salmonids near PCP treated utility poles. In 1991, the U.S. EPA determined that the use of pentachlorophenol poses the risk of oncogenicity because of the presence of hexachlorodibenzo-p-dioxin and hexachlorobenzene, both of which have the potential to produce teratogenic/fetotoxic effects) (CALEPA 1996).

Creosote is a rather complex chemical that is comprised of more than 160 different distillates that occur in coal-tar, including aromatic hydrocarbons (such as naphthalene, anthracene, benzene, toluene, xylene, acenaphthene, phenanthrene, and fluorene), tar acids (such as phenols, cresols, xylenols, and naphthols), and tar bases (including pyridines, guinolines, and acridines) many of which are toxicants and carcinogens (Shields 1976). The U.S. EPA determined that creosote has the potential for oncogenicity and mutagenicity (CALEPA 1996).

The following section describes a preliminary set of experiments conducted at UAB to investigate the potential of leaching of some of the different materials that can be used in construction.

5.0 Leaching of Various Construction Materials

Some construction material leaching tests were conducted by Pitt, *et al.* (2000) as part of a stormwater treatability research project. This project included the construction of pilot-scale treatment devices and there was concern about the selection of the construction materials that could affect the test results. Therefore, before the pilot-scale devices were constructed, as series of tests were conducted to examine the pollutant leachability of different potential construction materials. Samples of the various materials were left to soak in de-ionized water for set periods of time, and then the water was analyzed for a broad list of constituents of interest.

Table 5 lists potential contaminants from some materials that may be used in bench-scale and pilot-scale test equipment (Cowgill 1988). Cowgill found that extensive steam cleaning (at least 5 washings using steam produced from distilled water) practically eliminated all contamination problems for sampling equipment. Cemented materials should probably be avoided, as is evident from this table. Threaded or bolted together components are much preferable.

Table 5. Potential Sample Contamination from Sampler Material (Cowgill 1988)

| Material: | Contaminant: |
|--|---|
| PVC - threaded joints | chloroform |
| PVC - cemented joints | methylethyl ketone, toluene, acetone, methylene chloride, benzene, ethyl acetate, tetrahydrofuran, cyclohexanone, organic tin compounds, and vinyl chloride |
| Teflon™ | nothing |
| polypropylene and polyethylene | plasticizers and phthalates |
| fiberglass reinforced epoxy material (FRE) | nothing |
| stainless steel | chromium, iron, nickel, and molybdenum |
| glass | boron and silica |

Pitt, *et al.* (2000) tested the leaching potentials for many materials that may be used in bench-scale and pilot-scale treatment units, and some of these materials are likely exposed to stormwater during typical construction applications. Samples of each material were immersed for a period of 72 h in approximately 500 mL of laboratory grade 18 megohm water. A sample blank was also prepared. Analyses conducted on each of these samples, and the sample blank, were the same as performed for the pilot-scale treatment devices. Tables 6 and 7 present the contaminants that were found in the leaching water at the end of the test in high concentrations that may affect the test results. The most serious problems occur with plywood, including both treated and untreated wood. Attempting to seal the wood with Formica and caulking was partially successful, but toxicants were still leached. Covering of the Formica clad plywood with polyethylene plastic sheeting was finally used to eliminate any potential problem, for example. Fiberglass screening material, especially before cleaning, also causes a potential problem with plasticizers

and other organics. PVC and aluminum may be acceptable materials, if phthalate esters and aluminum contamination can be tolerated. The most serious concern is associated with the use of galvanized metals, as expected, where the tests indicated extremely high zinc concentrations, or the exposure of treated woods to stormwater (its typical application).

Table 6. Potential Sample Contamination from Construction Materials (Pitt, et al. 2000)

| Material: | Contaminant observed: |
|---|---|
| untreated plywood | toxicity, chloride, sulfate, sodium, potassium, calcium, 2,4-dimethylphenol, benzylbutyl phthalate, bis(2-ethylhexyl) phthalate, phenol, N-nitro-so-di-n-propylamine, 4-chloro-3-methylphenol, 2,4-dinitrotoluene, 4-nitrophenol, alpha BHC, gamma BHC, 4,4'-DDE, endosulfan II, methoxychlor, and endrin ketone |
| treated plywood (CCA) | toxicity, chloride, sulfate, sodium, potassium, hexachloroethane, 2,4-dimethylphenol, bis(2-chloroethoxy) methane, 2,4-dichlorophenol, benzylbutyl phthalate, bis(2-ethylhexyl) phthalate, phenol, 4-chloro-3-methylphenol, acenaphthene, 2,4-dinitrotoluene, 4-nitrophenol, alpha BHC, gamma BHC, beta BHC, 4,4'-DDE, 4,4'-DDD, endosulfan II, endosulfan sulfate, methoxychlor, endrin ketone, and copper (likely), chromium (likely), arsenic (likely) |
| treated plywood (CCA) and Formica | toxicity, chloride, sulfate, sodium, potassium, bis(2-chloroethyl) ether* , diethylphthalate, phenanthrene, anthracene, benzylbutyl phthalate, bis(2-ethylhexyl) phthalate, phenol* , N-nitro-so-di-n-propylamine, 4-chloro-3-methylphenol* , 4-nitrophenol, pentachlorophenol, alpha BHC, 4,4'-DDE, endosulfan II, methoxychlor, endrin ketone, and copper (likely), chromium (likely), arsenic (likely) |
| treated plywood (CCA), Formica and silica caulk | lowered pH, toxicity, bis(2-chloroethyl) ether* , hexachlorocyclopentadiene, diethylphthalate, bis(2-ethylhexyl) phthalate, phenol* , N-nitro-so-di-n-propylamine, 4-chloro-3-methylphenol* , alpha BHC, heptachlor epoxide, 4,4'-DDE, endosulfan II, and copper (likely), chromium (likely), arsenic (likely) |
| Formica and silica caulk | lowered pH, toxicity, 4-chloro-3-methylphenol, aldrin, and endosulfan 1 |
| silica caulk | lowered pH, toxicity, and heptachlor epoxide |
| PVC pipe | N-nitrosodiphenylamine, and 2,4-dinitrotoluene |
| PVC pipe with cemented joint | bis(2-ethylhexyl) phthalate* , acenaphthene, and endosulfan sulfate |
| plexiglass and plexiglass cement | naphthalene, benzylbutyl phthalate, and bis(2-ethylhexyl) phthalate, and endosulfan II |
| aluminum | toxicity, and aluminum (likely) |
| plastic aeration balls | 2,6-dinitrotoluene |
| filter fabric material | acenaphthylene, diethylphthalate, benzylbutyl phthalate, bis(2-ethylhexyl) phthalate, and pentachlorophenol |
| sorbent pillows | diethylphthalate, and bis(2-ethylhexyl) phthalate |
| black plastic fittings | pentachlorophenol |
| reinforced PVC tubing | diethylphthalate, and benzylbutyl phthalate |
| fiberglass window screening | toxicity, dimethylphthalate, diethylphthalate* , bis(2-ethylhexyl) phthalate, di-n-octyl phthalate, phenol, 4-nitrophenol, pentachlorophenol, and 4,4'-DDD |
| Delrin™ | benzylbutyl phthalate |
| Teflon™ | nothing (likely) |
| glass | zinc (likely) |

note: * signifies that the observed concentrations in the leaching solution were very large compared to the other materials.

Table 7. Analyses of Washoff from Various Construction Materials

| Sample | Copper (µg/L) | Cadmium (µg/L) | Lead (µg/L) | Zinc (µg/L) | Iron (µg/L) | Chromium (µg/L) | Magnesium (µg/L) | Calcium (µg/L) |
|---------------------------------|---------------|-------------------|-------------|-------------|-------------|-----------------|------------------|----------------|
| silica caulk | 29 | <lod ¹ | <lod | 14 | 48 | 8 | <lod | 0.08 |
| formica and silica caulk | 54 | <lod | <lod | 26 | 110 | 8 | <lod | 0.38 |
| metal roof runoff | 41 | <lod | 32 | 10,200 | 440 | 11 | 0.13 | 1.2 |
| treated plywood | 1,300 | <lod | 33 | 93 | 110 | 2,800 | 0.02 | 0.67 |
| untreated plywood | 79 | <lod | <lod | 67 | 310 | 12 | 1.3 | 3.2 |
| washed PVC and PVC cement | 36 | <lod | <lod | 32 | 83 | 8 | <lod | 0.60 |
| washed geotextile filter fabric | 44 | <lod | <lod | 32 | 110 | 16 | 0.05 | 1.2 |
| washed fiberglass window screen | 32 | 17 | <lod | 88 | 47 | 8 | <lod | 0.10 |

¹ <lod: less than the limit of detection.

Table 8 summarizes the selected materials used in the construction of the test apparatus. These tables indicate that care must be taken when selecting test equipment. The use of Teflon™ reduces most of the problems, but it is quite expensive. Delrin™ is almost as effective, is somewhat less expensive, and is much easier to machine when manufacturing custom equipment. Both of these materials are fragile and cannot withstand rough handling. Glass is not usable for most large treatability test equipment, but is commonly used in bench-scale tests.

Table 8. Preliminary Construction Material Leach Test (Pitt, et al. 2000)

| MATERIAL | LEACH POTENTIAL |
|--|--|
| PVC pipe and cement | LOW |
| polyethylene sheeting | LOW (n-nitroso-di-n-propylamine) |
| Plexiglas™ and cement | LOW (conductivity, chloride, sodium) |
| Formica™ and caulk | LOW (toxicity, conductivity, pH, nitrobenzene, 4-chloro-3-methylphenol) |
| aluminum angle bracket | LOW (toxicity, conductivity, chloride, calcium, pentachlorophenol) |
| Amoco 4557 filter fabric (Gunderboom™) | LOW (toxicity, conductivity, sulfate, pentachlorophenol) |
| plastic screen | HIGH (toxicity) |
| treated plywood | LOW (phenol, 4-nitrophenol, pentachlorophenol, di-n-octylphthalate) |
| | HIGH (toxicity, hexachloroethane, 2,4-dimethylphenol, 4-chloro-3-methylphenol, 4-nitrophenol; likely heavy metals) |

6.0 Conclusions

This paper presented information showing the potential benefits of using alternative building materials as a stormwater management control. As an example, although roof runoff may only contribute about 10 to 20 percent of all stormwater runoff from typical residential areas, almost all of the zinc has been found to originate from this source due to the use of galvanized metal roof flashing and drainage gutters and downspouts. Serious problems may also be associated with the use of other metals on buildings, especially copper. Pavement contributes large fractions of the total stormwater runoff volume in most areas, and the selection of different pavement materials may have significant effects on runoff quality, although there is currently very little supportive data. The use of other building materials, especially treated wood, may also have significant adverse effects on runoff quality. It is likely that

Careful selection of building materials may help reduce stormwater pollutant concentrations, although additional research is needed to quantify the likely benefits and to test different materials and associated pollutant release and fate processes.

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